

Hi! Thank you for your download. I'm so glad you were able to find a school tool you can use. Please feel free to use this activity for your own personal use or classroom. Hope it works out great!



♥ Marie



- **TERMS OF USE:** THIS DOCUMENT IS PROVIDED TO YOU FOR YOUR OWN PERSONAL USE. YOU AGREE THAT YOU WILL NOT COPY, REPRODUCE, ALTER, MODIFY, CREATE DERIVATIVE WORKS, OR PUBLICLY DISPLAY CONTENTS AS YOUR OWN. NO REDISTRIBUTION. YOU MAY NOT REPRODUCE, REPACKAGE, OR REDISTRIBUTE THE CONTENTS OF THESE DOWNLOADS, IN WHOLE OR IN PART, FOR ANY REASON. THIS INCLUDES "GIVING" SOMEONE YOUR COPY THAT YOU ARE NO LONGER USING OR HOSTING THEM ON DROP BOX OR FACEBOOK FILES. PLEASE REFER OTHERS TO WWW.THEHOMESCHOOLDAILY.COM TO DOWNLOAD THEIR OWN COPY.

YOU MAY:

- Save the files on your computer and print off copies for your family or classroom whenever you would like.
- Link directly to my blog to share my files with others.
- Post to your blog using pictures of your child using my curriculum, as long as proper credit is given to www.thehomeschooldaily.com

YOU MAY NOT:

- Host or store my files on your own or other sites (this includes drop box, the cloud, and any other site off your personal computer)
- Alter or Sell files to make a profit. All files are for personal/classroom use only.
- All downloads are copyright protected. Not to be distributed, transferred, or shared in any form.



Candy Bar Density Lab

Find out which candy bars will “satisfy” your hunger! Snickers claims to “satisfy,” but what about 3 Musketeers?

If Density is the amount of mass in volume, then wouldn't the candy bar with the greatest density satisfy your hunger the best? So, which candy bar has the highest density? Let's investigate!



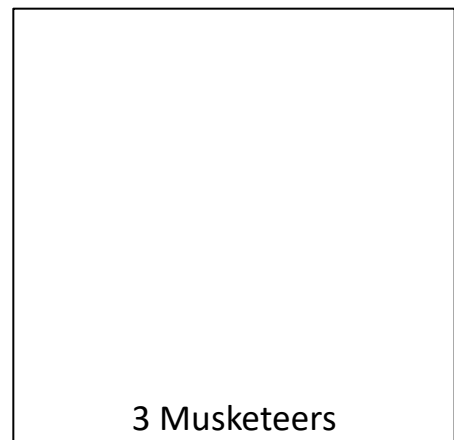
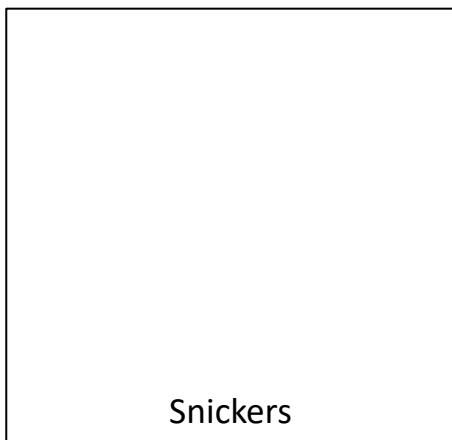
We are going to make observations about each candy bar, find the mass, volume, and density of each candy bar, and then test their density against water.

• Gather Materials

- 2 Paper Plates
- Knife
- 2 Snickers
- 2 3 Musketeers
- Ruler
- Triple Beam Balance
- Calculator
- Beaker filled with water $\frac{1}{2}$ to $\frac{3}{4}$ full
- Paper Towel or Napkins
- Lab Sheet

• Make Observations

- Open one Snickers and one 3 Musketeers. (Do not eat!) Put each of them on their own paper plate.
- Using the knife, cut the candy bars in half.
- Illustrate what each looks like. Write 3 words that describe what the contents look like.



- Based on what you observe about each candy bar, which one do you think is denser? Why? _____
-

- Collect and Record Data**

- 1) Open the remaining two candy bars: one 3 Musketeers and one Snickers. Put them on their corresponding paper plates.
- 2) Now, using a triple beam balance, find the mass of each candy bar and then record the data in the table below.
- 3) Next, using a ruler, find the length, width, and height of each candy bar in centimeters. Record your data in the table below.
- 4) Next, calculate each candy bar's volume. To do this, you will multiply its length x width x height. Use the calculator and then record your data.
- 5) Now that you have the mass and volume of each candy bar, find their densities. You will do this by dividing. Here is the formula: $\text{MASS} / \text{VOLUME}$. Use your calculator to keep it simple.

	Mass in Grams (g)	Length measured in centimeters	Width measured in centimeters	Height measured in centimeters	Volume in centimeters (cm ³) L x W x H = V	Density Mass/Volume g/cm ³
3 Musketeers						
Snickers						

- **Facts to Consider**

- Before we continue, let's discuss buoyancy and the density of water.

Buoyancy is the *ability of an object to float in water or air*. The density of pure water is always 1.0g/cm^3 . An object with a density of more than 1.0g/cm^3 will sink and an object with a density of less than 1.0g/cm^3 will float.

- **Analyze Data to Make Hypothesis**

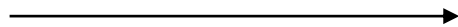
- If I put a 3 Musketeer candy bar into water, then it will _____.
- If I put a Snickers candy bar into water, then it will _____.

- **Test Hypotheses**

- Drop each candy bar into the water.

- **Draw Conclusions**

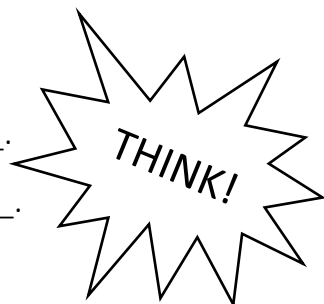
- Draw what happened in your experiment.



- Was your hypotheses correct? Explain. _____

- Fill in the Blanks with either the word increase or decrease.

- When you increase mass, density will _____.
- When you increase volume, density will _____.
- When you decrease volume, density will _____.
- When you decrease mass, density will _____.



* Did you know that Earth is the densest planet in the solar system? Saturn is the least dense. If Saturn was placed into a large tub of water, it would float. If Earth was put in a tub of water, it would sink.